

ICP - Mass Spectrometry

APPLICATION NOTE

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Determination of CeO₂ Nanoparticle Plant Uptake using Single Particle ICP-MS

Introduction

Cerium dioxide (CeO₂) nanoparticles have found wide use in industrial and commercial products, such as coating materials, fuel additives,

and nanomedicine, to name a few. As such, it is expected that CeO_2 nanoparticles (NPs) will find their way into the environment and may be transported via aqueous systems where they may end up in drinking water and/or plants, both of which increase the possibility of consumption by humans. It has been shown that drinking water treatment removes CeO_2 NPs¹, so the likelihood of ingestion by humans is reduced. However, several studies have demonstrated plant uptake of CeO_2 NPs, although the mechanism is unknown²⁻⁴.

It has been demonstrated that single particle ICP-MS (SP-ICP-MS) can effectively detect nanoparticles in plant tissue⁴ and other biological systems⁵. The benefits of SP-ICP-MS for these types of studies include the ability to rapidly count and size NPs, even at low concentrations, as well as determine the dissolved metal content. SP-ICP-MS has been shown to effectively measure these NPs in several different matrices.



The goals of this work are to develop a SP-ICP-MS method and to characterize the uptake and accumulation of CeO₂ NPs in several different agricultural crops. The detailed results and discussion of this work have been published recently⁶, so only an overview will be given here.

Experimental

Materials and Samples

Two different sizes of CeO₂ NPs were used in this study: 30-50 nm (US Research Nanomaterials™, Houston, Texas, USA) and 50-100 nm (Nanostructured and Amorphous Materials Inc., Houston, Texas, USA). In addition, citrate-stabilized 50 nm gold NPs (nanoComposix™, San Diego, California, USA) were used to determine the transport efficiency of the SP-ICP-MS system. Other materials used included tomato, cucumber, pumpkin, and soybean seeds (Johnny's Selected Seeds, Winslow, Maine, USA), Macroenzyme R-10 (bioWORLD™, Dublin, Ohio, USA) for enzymatic digestion of the plants, 1000 mg/L Ce standard (High Purity Standards™, Charleston, South Carolina, USA), and 2-(N-morpholino) ethanesulfonic acid (MES; Sigma-Aldrich™, St. Louis, Missouri, USA).

Plant Growth and CeO₂ Nanoparticle Exposure

Plants were grown from seeds under controlled conditions at $25\,^{\circ}\text{C}$ with 16 hours of light and eight hours of darkness per day. After 1.5-2.5 weeks, 7 mg/L of the 30-50 nm CeO₂ NPs were added to the growing medium. The plants were then harvested for analysis after 19 days of exposure to the CeO₂ NPs. Three replicate plants of each species, as well as controls (i.e. plants without CeO₂ NPs dosed), were grown.

Enzymatic Digestion of Plant Shoots

For NP analysis, it is important to extract the NPs without dissolving them. For that to be accomplished, traditional acidic digestions cannot be used; instead, a recently developed enzymatic digestion method⁴ was used with slight modification. After harvesting, the shoots were separated from roots, washed, cut into small pieces, and homogenized in 20 mM MES at pH 5. Next, the enzyme solution (30 mg/mL of Macerozyme R-10 in 20 mM MES) was added, and the samples were agitated for 24 hours at 37 °C in a waterbath shaker. The samples were allowed to settle for 30-60 minutes before 0.1 mL of the supernatant was removed and diluted 100-fold with the MES for SP-ICP-MS analysis.

Instrumentation and Analysis

All analyses were done on a PerkinElmer NexION® 350D ICP-MS using the conditions shown in Table 1. For SP-ICP-MS, all data collection and analysis was performed using the Syngistix™ Nano Application Software Module. Ce was monitored at m/z 140 for all measurements due to its high natural abundance (88.48%) and lack of interferences. Dissolved Ce prepared in diluted plant matrix was used to calibrate the particulate Ce as CeO₂ and to quantify the dissolved Ce in plant shoots. The transport efficiency was determined using gold NPs in plant shoot matrices.

Table 1. NexION 350D ICP-MS Instrumental and Method Conditions.

Instrumental Conditions and Parameters		
Nebulizer	MEINHARD Type C, glass	
Spray Chamber	Cyclonic, glass	
Sample Uptake Rate (mL/min)	0.27	
RF Power (W)	1600	
Cones	Platinum	
SP-ICP-MS Method Parameters		
Analyte	Ce	
Mass (amu)	140	
Dwell Time (ms)	0.1	
Sampling Time (s)	100	
CeO ₂ Density (g/cm³)	7.13	
Ce Mass Fraction in CeO ₂ (%)	81.39	

Results and Discussion

The accuracy of the developed SP-ICP-MS method was validated using two commercial CeO_2 NPs (also characterized by transmission electron microscope [TEM]). The 30-50 nm and 50-100 nm CeO_2 NPs were prepared in the 20 mM MES buffer to match the matrix used in the plant uptake experiments. Figures 1(a) and 1(c) show the particle size distributions measured by SP-ICP-MS. The size measured by SP-ICP-MS is comparable to the size measured by TEM images (Figures 1(b) and 1(d)). The long tail on the plots indicates that the particles contain a wide range of sizes, which was confirmed by TEM. These results also indicate that the MES buffer does not affect the particle size measurements.

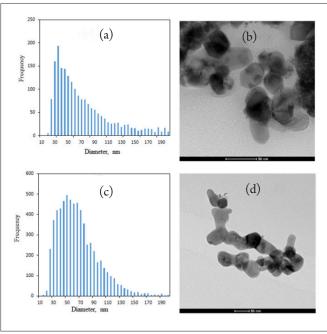


Figure 1. Comparison of size distributions of CeO $_2$ NPs in 20 mM MES by SP-ICP-MS and TEM imaging. (a) Histogram of 30-50 nm CeO $_2$ NPs measured by SP-ICP-MS; (b) TEM image of 30-50 nm CeO $_2$ NPs; (c) Histogram of 50-100 nm CeO $_2$ NPs measured by SP-ICP-MS; (d) TEM image of 50-100 nm CeO $_2$ NPs (scale bar on both TEM images is 50 nm).

After the accuracy of the methodology was established, the $50\text{-}100 \text{ nm CeO}_2$ NPs were spiked into the enzyme-digested shoot sample of each plant type after 100-fold dilution in the 20 mM MES buffer. The results of matching sizes indicate that the diluted plant matrix does not affect the SP-ICP-MS size measurements. One of the representative size distribution histograms is shown in Figure 2(a).

To explore the effect of the enzymatic digestion on CeO_2 NP size measurement, two solutions of the 1 µg/L, 50-100 nm CeO_2 NPs were made: one was measured immediately, while the other was digested with the enzyme for 24 hours, as described above. Figures 2(b) and 2(c) show the particle size distributions of three replicate measurements of each solution. Since the size distributions and measurements are the same for both solutions, it is established that enzyme digestion does not affect the CeO_2 NPs. In addition, good reproducibility of the SP-ICP-MS measurements is also established.

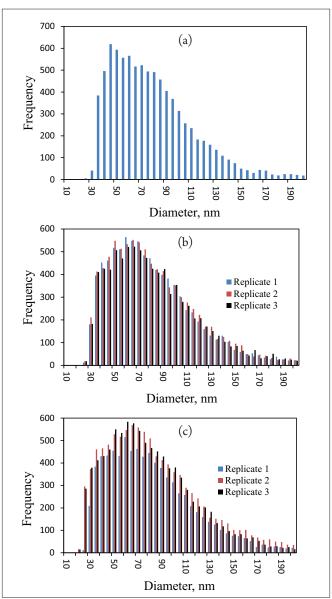


Figure 2. (a) Size distribution of the 50-100 nm CeO $_2$ NPs spiked in the tomato plant digestates after 100-fold dilution in the 20 mM MES buffer; (b) The size distribution of 50-100 nm CeO $_2$ NPs after 24-hour digestion in enzyme solution at 37 °C; (c) the size distribution of freshly prepared 50-100 nm CeO $_2$ NPs in enzyme solution.

To determine if CeO₂ NP uptake by plants is occurring, samples of each plant grown both with and without exposure to CeO₂ NPs were studied. The exposed plants were in contact with 7 mg/L of the 30-50 nm CeO₂ NPs for 19 days hydroponically. Figure 3 shows the raw data plots (time vs. intensity) of the results for tomato shoots. Similar results were obtained from the other plant shoot samples. The data in Figure 3(a) is for the plant shoots which were not exposed to the CeO₂ NPs, and the data in Figure 3(b) is for the plant shoots that had CeO₂ NP exposure. No significant particulate Ce or dissolved Ce was seen for the plant without CeO₂ NP exposure. The continuous signal shown in the Figure 3(b) indicates that dissolved Ce was present in the plant shoot after being exposed to CeO₂ NPs for 19 days hydroponically. Selected samples were also filtered through Millipore™ 5 kDa Ultrafree®-MC centrifugal filters after enzymatic digestion. Continuous signal was still observed, which confirmed the presence of dissolved Ce in the plant shoot, and subsequently indicates CeO₂ NPs had been biotransformed to dissolved Ce by the plants.

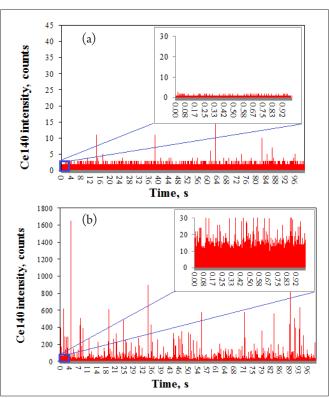


Figure 3. (a) Raw data of plant tomato shoots not exposed to CeO_2 NPs; (b) Raw data of plant tomato shoots exposed to 30-50 nm CeO_2 NPs (7 mg/L) for 19 days. The insets show the data points for the first second of data acquisition.

The pulse signals in Figure 3(b) represent particulate Ce. These pulsed signals might be either CeO_2 NPs, undigested plant tissue residual colloids with Ce ions adsorbed onto them, or a combination of both. The size distribution of particulate Ce shown in Figure 4 suggests dosed CeO_2 NPs. However, this could not be confirmed by SP-ICP-MS, and further investigation is needed. Nevertheless, this study unequivocally demonstrated the presence of dissolved Ce in plant shoots hydroponically exposed to CeO_2 NPs for the first time. In addition, the unique capability of the SP-ICP-MS methodology allows the dissolved and particulate forms of NPs in plants to be distinguished.

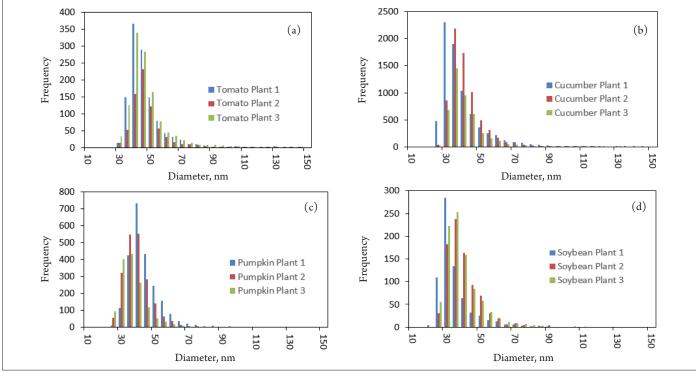


Figure 4. Size distribution histograms of particulate Ce (processed as CeO₂NPs) in plant shoots dosed with 7 mg/L 30-50 nm CeO₂NPs for 19 days: (a) tomato, (b) cucumber, (c) pumpkin, (d) soybean.

Conclusion

With the prevalence of nanoparticles, their migration into the environment is inevitable, creating the need to understand the interaction of NPs with environmental components. This work has developed a SP-ICP-MS method for simultaneous detection of CeO₂ NPs size and size distribution, particle concentration, and dissolved Ce in plant tissues. The method has been used to explore the uptake of CeO₂ NPs by a variety of plant species. By monitoring the digested plant components with SP-ICP-MS, both particulate Ce and dissolved Ce were reported in plant tissues for the first time, demonstrating the unique capability of SP-ICP-MS technology.

References

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Consumables Used

Component	Description	Part Number
Gold Standard	1000 ppm, 125 mL	N9303806
Gold Nanoparticles	50 nm, 4.5E+10 particles/mL, 25 mL	N8142302
Sample Uptake Tubing	0.38 mm id (green/orange), PVC, flared, 2 stop	N0777042
Drain Tubing	1.30 mm id (gray/gray), Santoprene, 2 stop	N0777444

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